

Effect of MoO₃ doping and grain size on SnO₂-enhancement of sensitivity and selectivity for CO and H₂

Abstract

MoO₃-doped SnO₂; nano-particles were prepared using sol-gel method with 0.1 molar solutions of SnCl₄·5H₂O and MoO₃ with different doping concentration from 5.03 × 10⁻⁴ to 5.04 × 10⁻³. The average grain size varies from 12 to 80 nm obtained by varying calcinations temperature from 300 to 900 °C. Thick film samples, of about 15 μm thickness, were tested for CO and H₂ in the concentration range of 15-1000 ppm in ambient air. The samples exhibit selective response to CO and H₂ at 140 and 240 °C, respectively. The sensitivity increases with doping concentration for H₂. Whereas, for CO, it increases rapidly till the doping concentration of 2.01 × 10⁻³ mol and then reaches to saturation. The observed maximum sensitivity factor for CO at 2.01 × 10⁻³ mol doping is ~30, whereas for H₂, it is 10 at 5.04 × 10⁻³ mol doping. The response and recovery time is ~5 and 10 s respectively.

1. Introduction

Over the past few decades, solid state sensors based on semi-conducting oxides and composites are the subject of study for detection of various hazardous gases like H₂ [1-3], NO_x [4], H₂S [5] and CO [6-13], etc. The gas sensing mechanism involves the redox reaction at the surface, leading to the changes in the depletion layer in the grain and hence changes in the electrical resistance [1,2,6-8]. Although such sensors are in daily use, their absolute stability is still poor susceptible to a high degree of drift even with slight change in the ambient conditions [1,6-8]. The important factor, which is still to be addressed, is the selectivity of the sensors towards a particular gas or species [1-14]. The main disadvantage of the commercially available sensors is the high temperature operation due to which the power consumption is of main concern.

In the group of hazardous gases, CO is being focused due to its serious toxicity with no smell and hence attaining the importance to control at sub-ppm level. An extensive research is being carried out using various metal

oxides like ZnO, In₂O₃, MoO₃, SnO₂ as smart sensor materials [1] for better sensor performance. Amongst these, SnO₂ is the most studied and suitable candidate as it represents a good compromise between cost (economy) and reliability along with comparatively low operating temperature with a disadvantage of instability and sensitivity to various gases [1-6, 11-20]. Therefore, still this material needs to be studied to improve the lacking features. It has been so far demonstrated that the sensing characteristics of SnO₂ based sensors can be improved by considering several factors like decreasing of crystallite size. The composition and valency control, addition of some noble metals as catalyst like Pt, Pd, etc. or addition of various metal oxides [1-5, 8, 12-16]. Relatively small efforts have been done to study the effect of MoO₃ with the earlier mentioned metal oxide in comparison with noble metals [10]. In the present work. An attempt is made to improve the SnO₂ sensor performance with MoO₃ doping for CO and H₂ gases. As the starting point, calcination temperature and the doping concentrations are varied systematically and also effect of grain size on the sensitivity is also carried out.